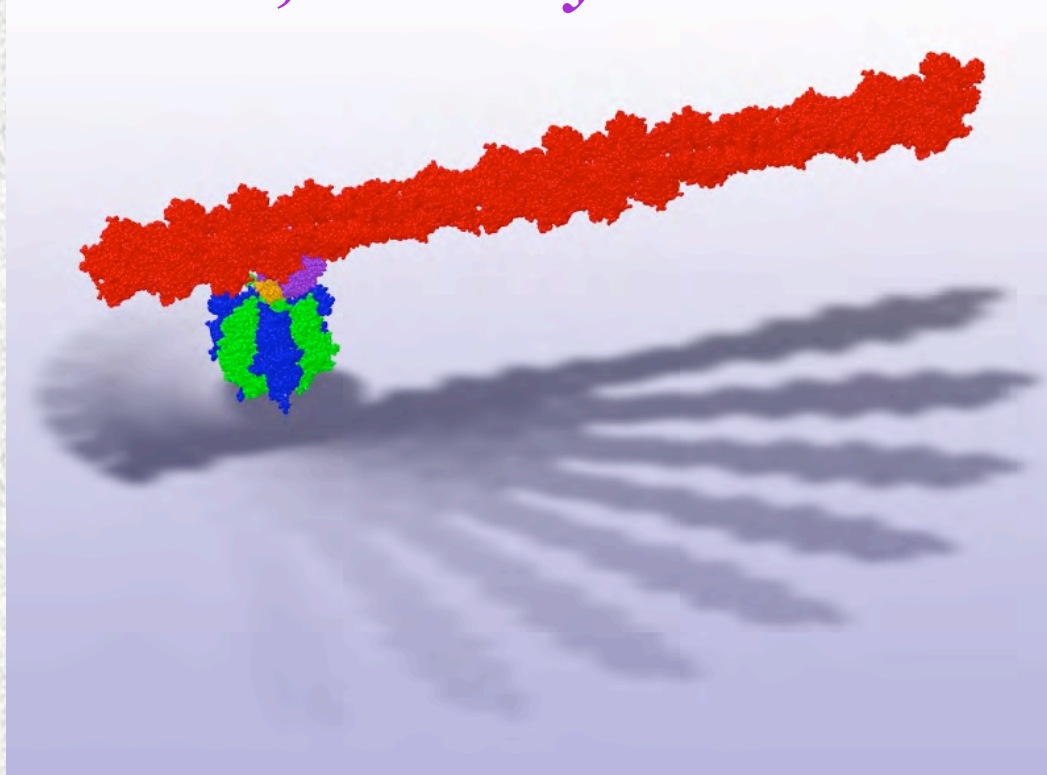


F_1 - ATPase: A Molecular Transducer of Chemical and Mechanical Energies

Why Do We Eat, and Why Do We Breathe?

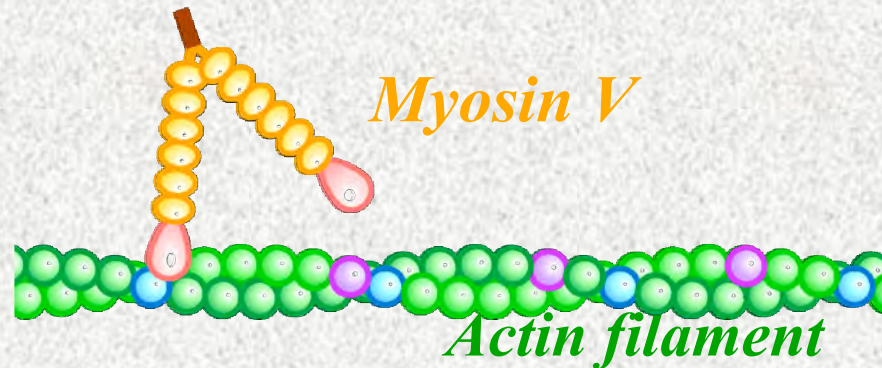


Kazuhiko Kinosita, Jr.

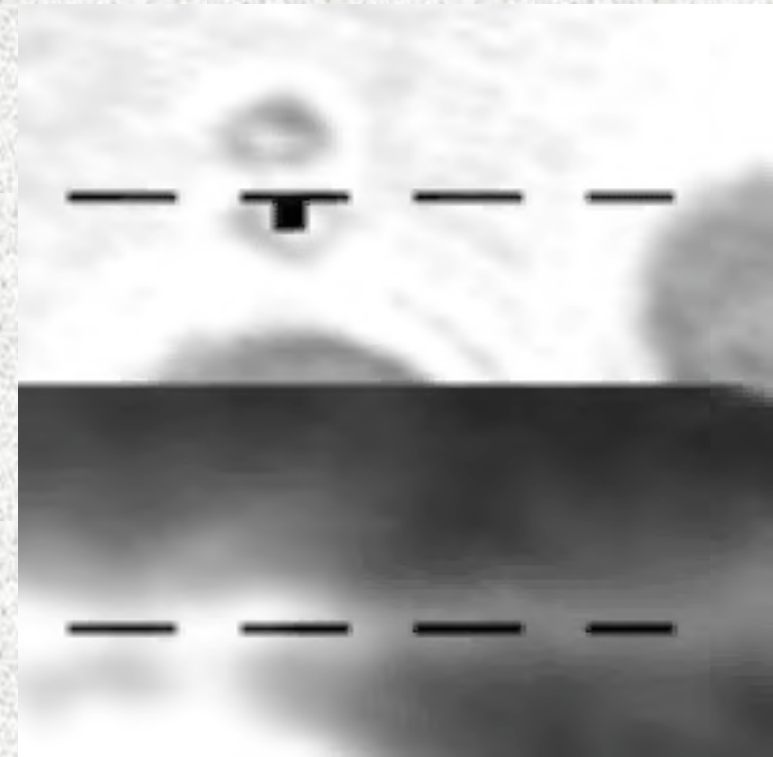
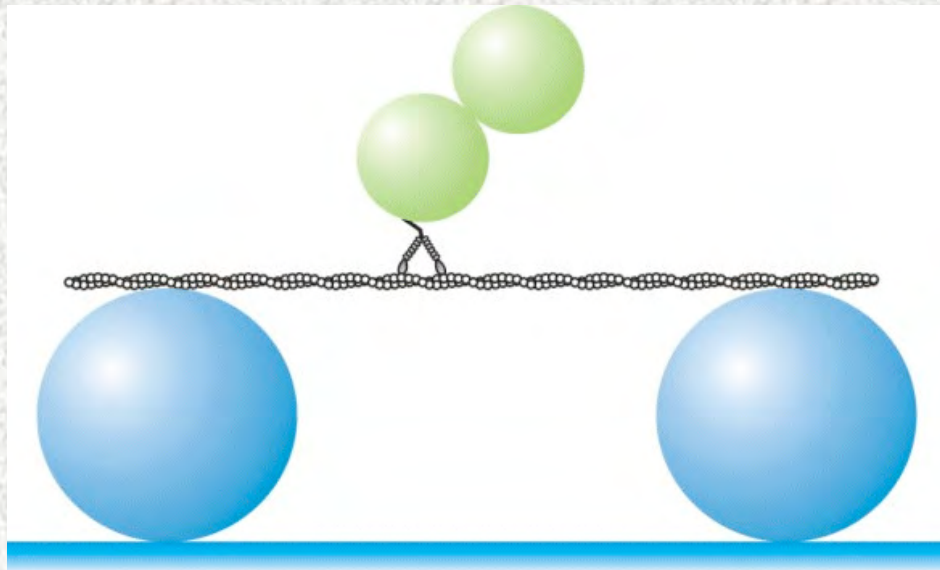
*Department of Physics, School of Science and Engineering
Waseda University*

<http://www.k2.phys.waseda.ac.jp>

A Two-Foot Linear Molecular Motor That Transports Nutrients in the Brain



White: closer to you
Black: away from you
4x fast replay



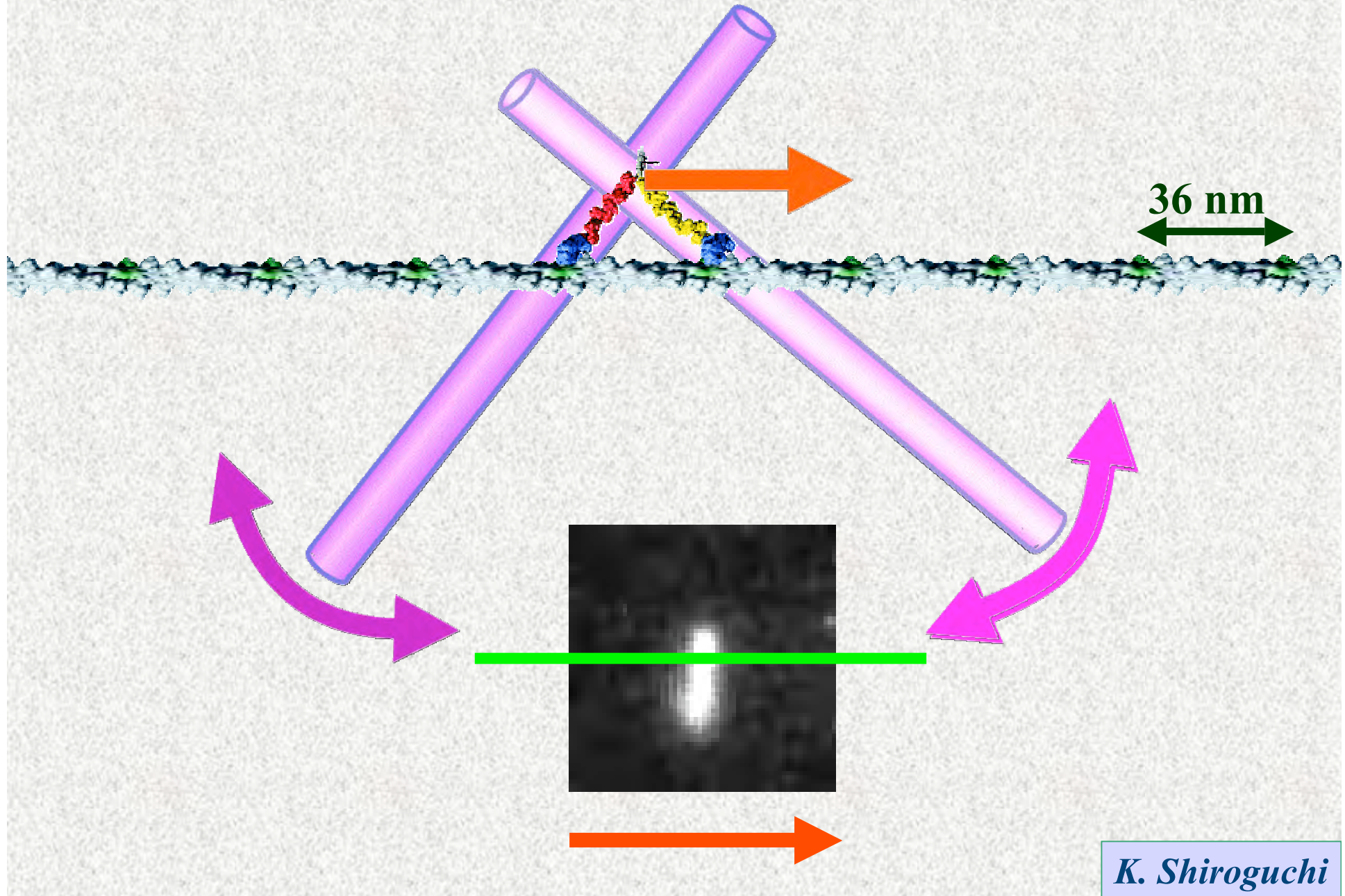
Major Players

400 μ M ATP

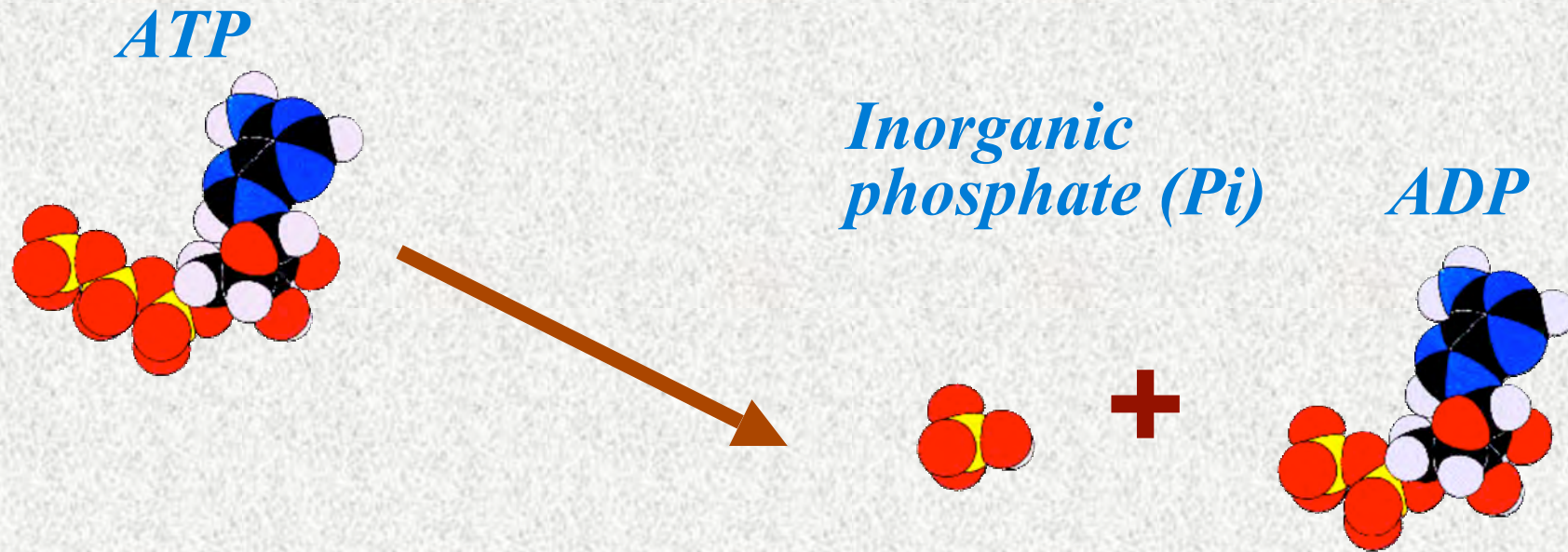
Literature quotation

M. Y. Ali, S. Uemura, S. Ishiwata

Watching How Myosin May Walk



ATP (Adenosine Triphosphate): A Universal Currency of Energy in the Biological World



Free energy drop:

$$\Delta G = \Delta G_0 + k_B T \ln[ADP][Pi]/[ATP]$$

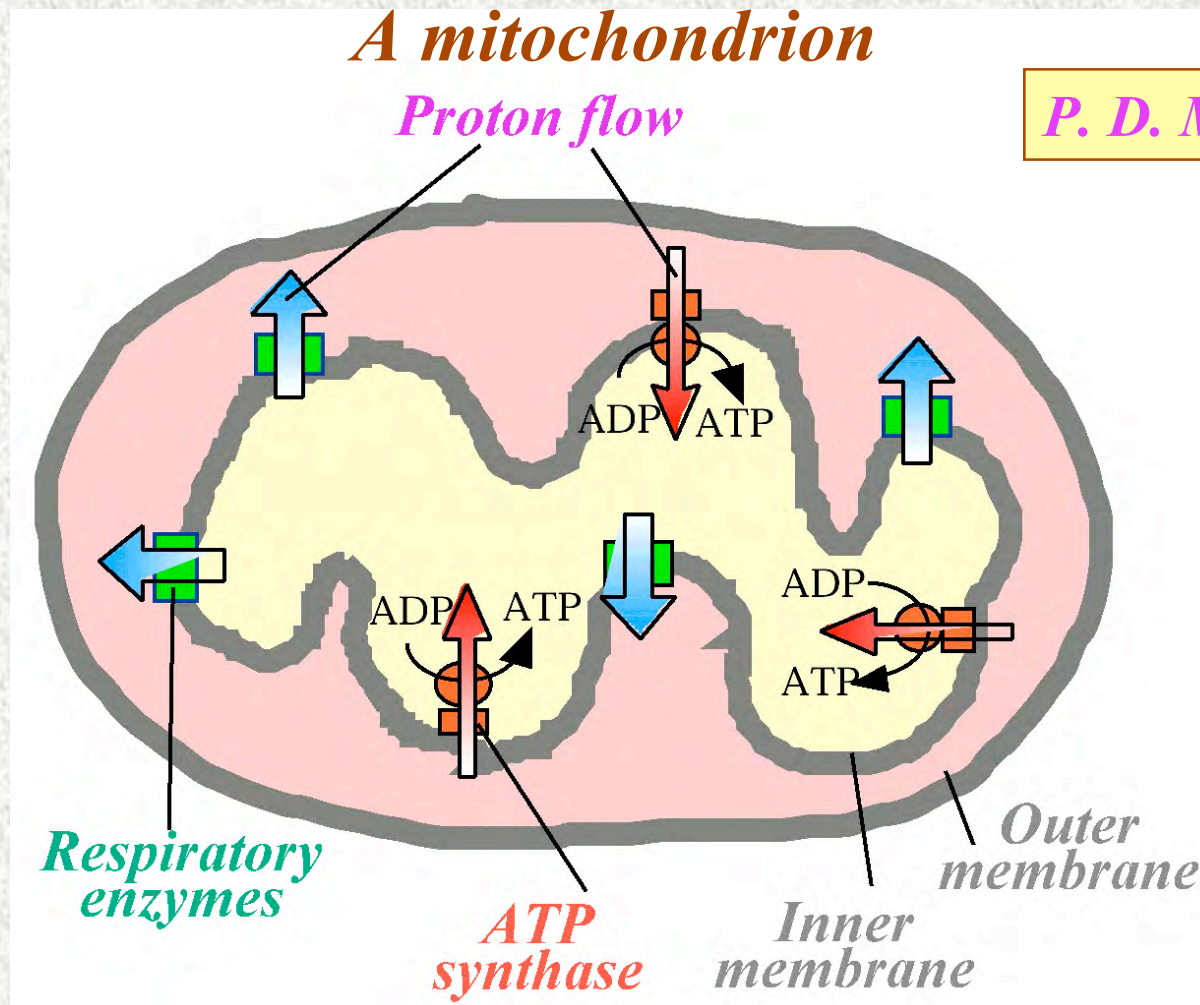
$$\Delta G_0 = -50 \text{ pN}\cdot\text{nm} \quad (@\text{pH } 7)$$

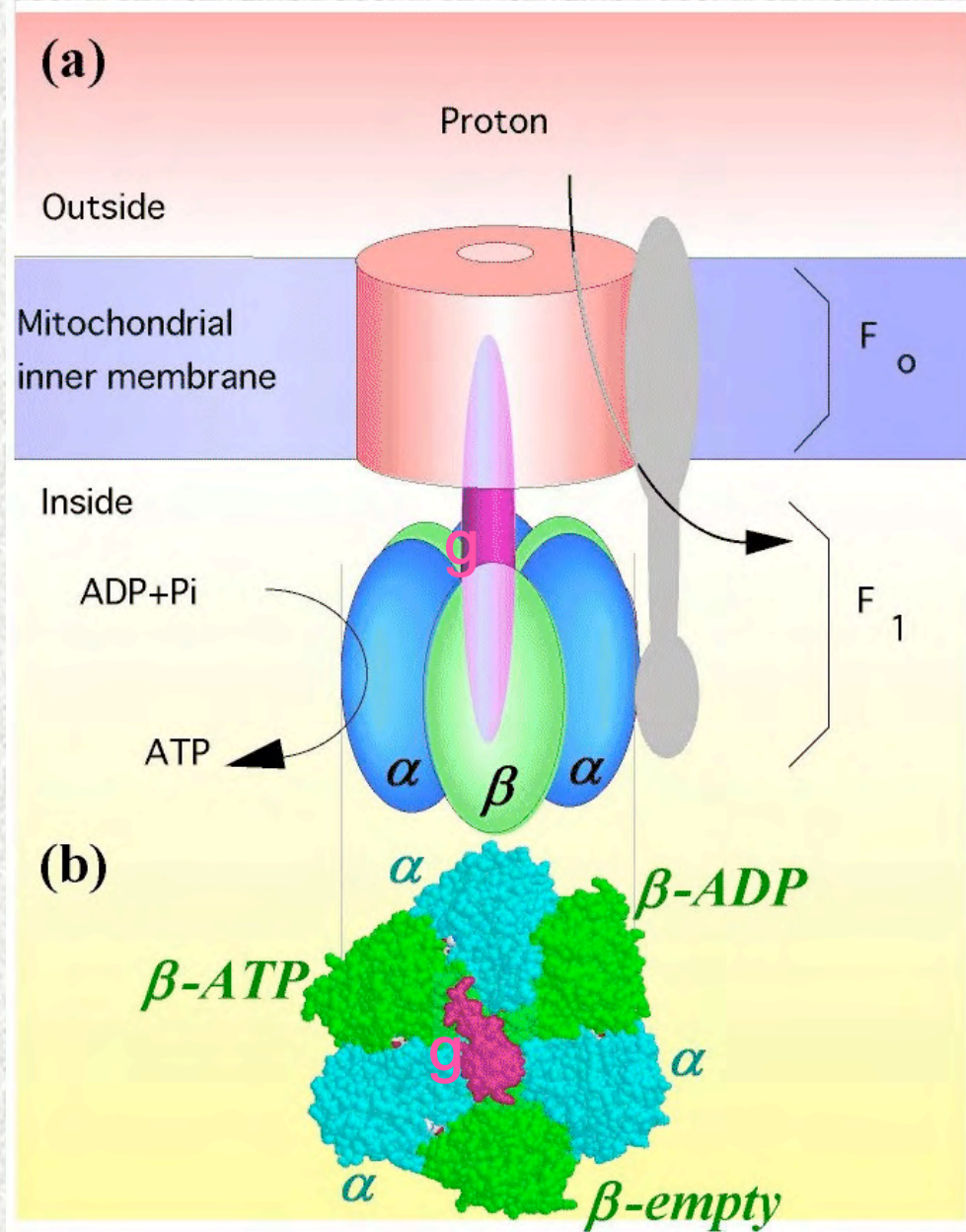
$$\Delta G = -(80\sim 100) \text{ pN}\cdot\text{nm} \text{ in cells}$$

$$= -(20\sim 25) k_B T$$

Why do We Eat, and Why do We Breathe?

*To obtain energy (= to synthesize ATP)
by slowly burning the food we ingest with oxygen we inhale.*

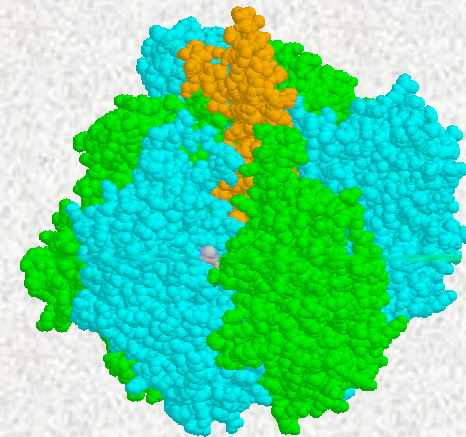




ATP Synthase

Proton-driven and ATP-driven motors connected by a common shaft (γ).

ATP synthesis/hydrolysis on three β s.



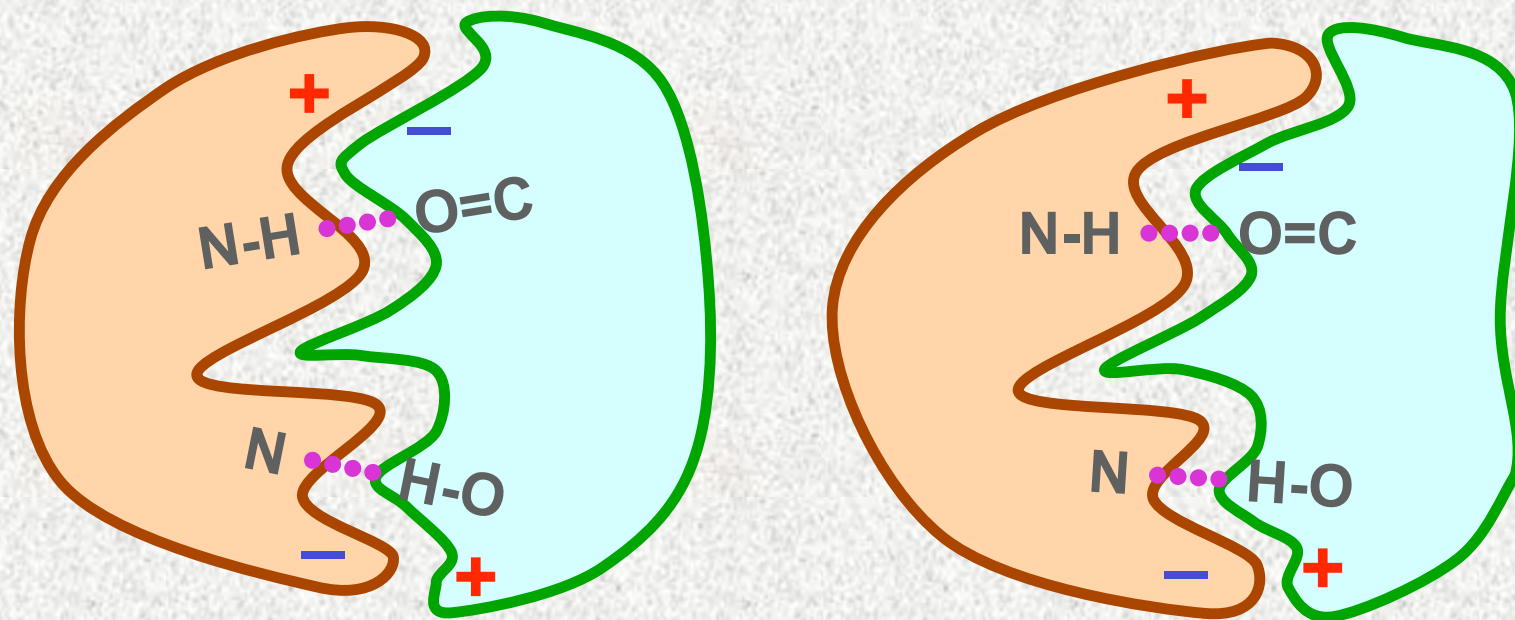
Isolated F_1 only hydrolyzes ATP; hence it is called F_1 -ATPase

D. Boyer

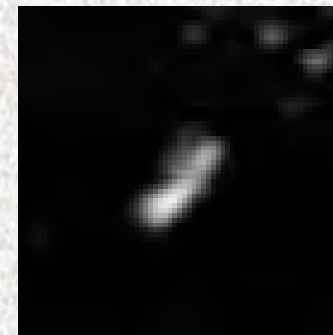
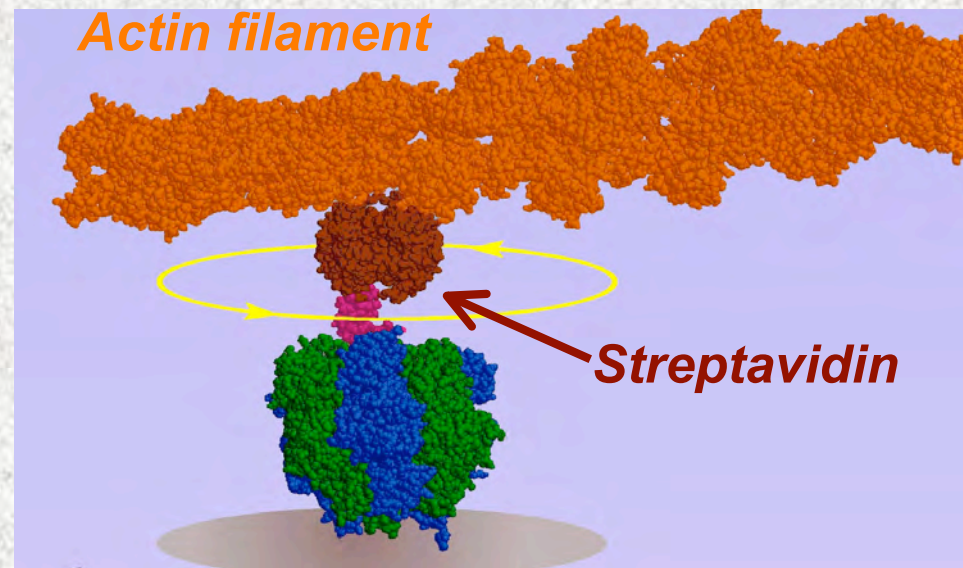
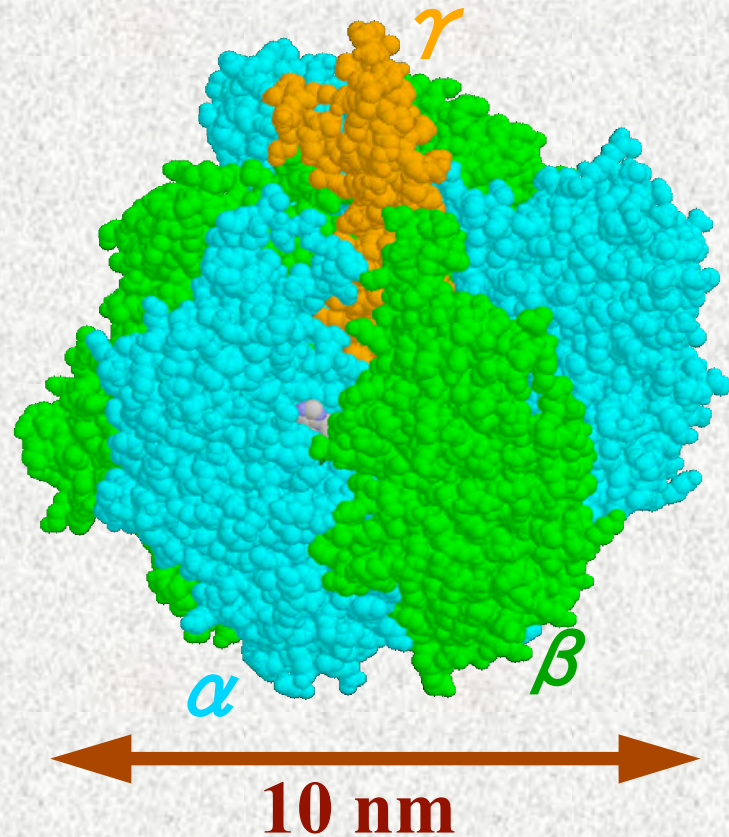
F. Oosawa & S. Hayashi

J. E. Walker & colleagues

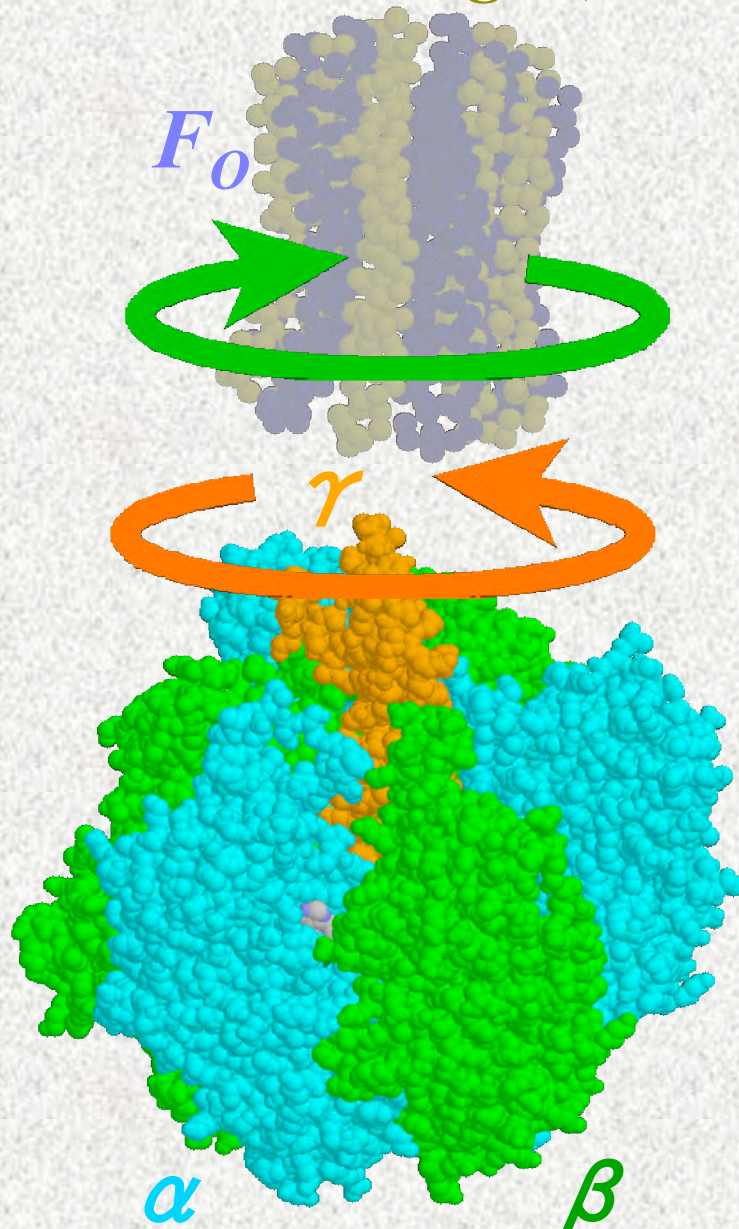
*A Protein Molecule Is Made
by Lock-and-Key Mechanisms*



F_1 -ATPase: A Rotary Motor Made of a Single Molecule



*Physiological Function of F_1 Is to Synthesize ATP
through (Forced) Clockwise Rotation*



*Isolated F_1 :
CCW rotation powered by
ATP hydrolysis
in the three β subunits*

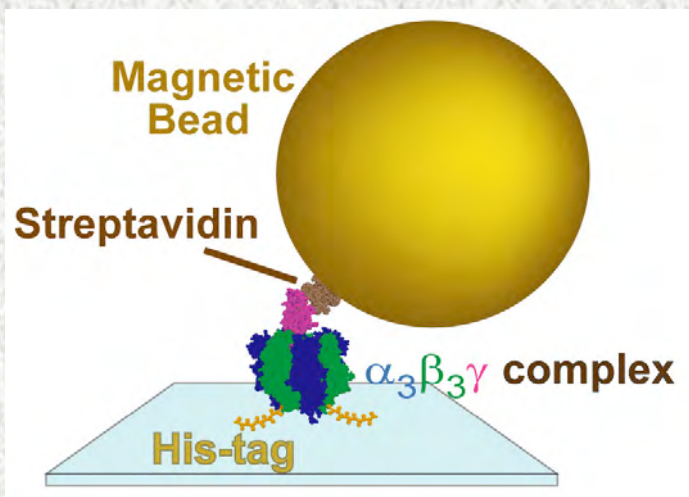
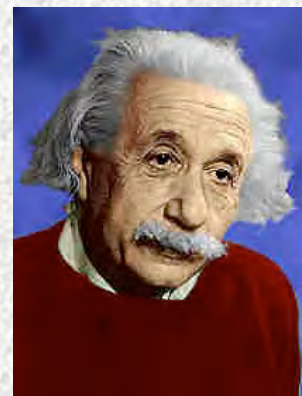


*F_1 in vivo:
CW rotation driven by proton
flow through the F_0 motor*

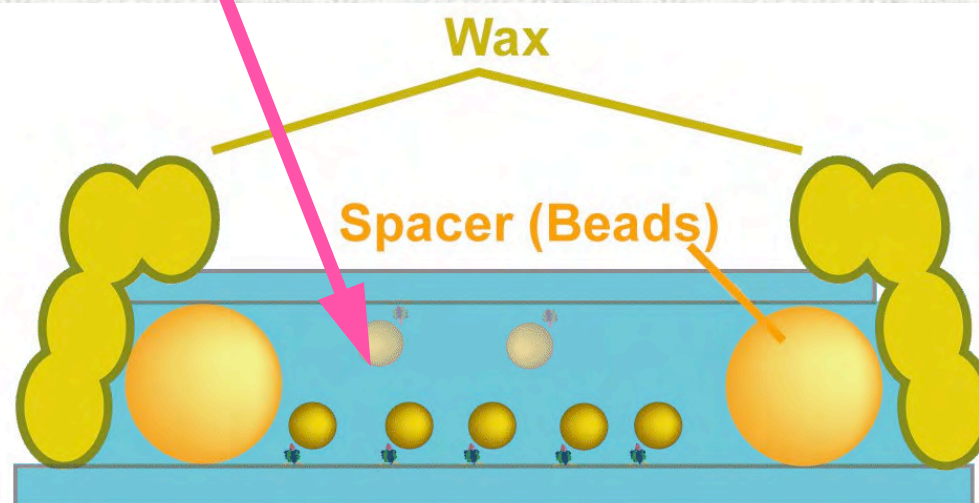
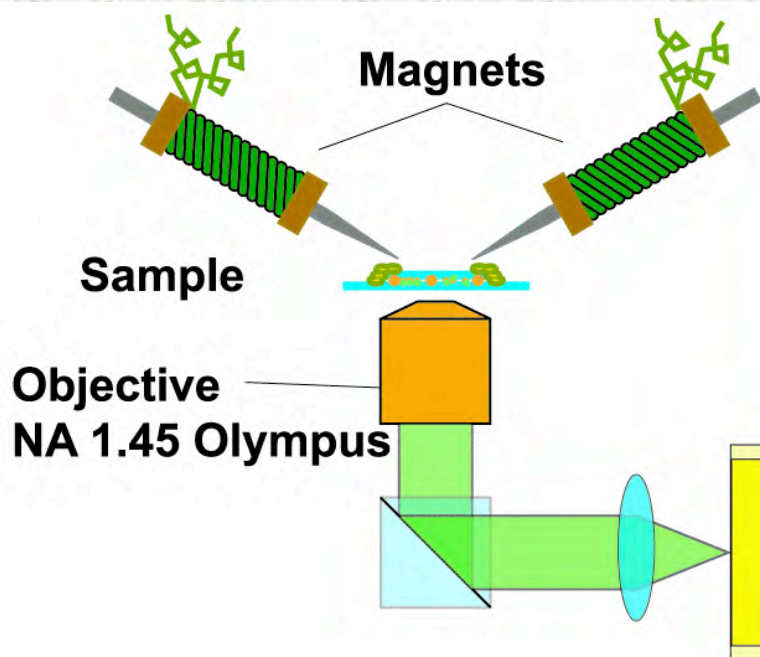


*ATP synthesis
in the three β subunits*

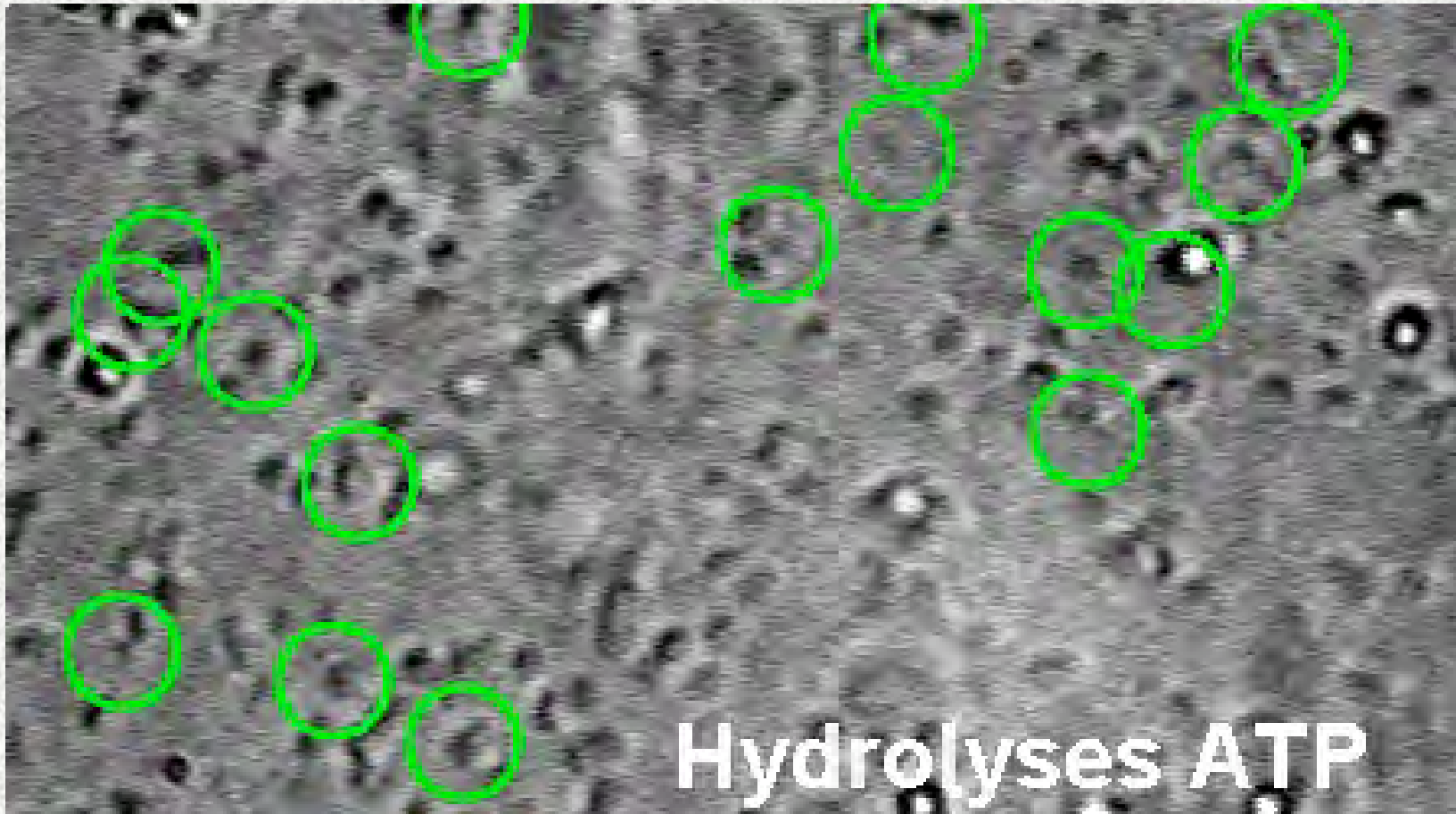
Attempt at Mechanical Synthesis of ATP



Luciferin/Luciferase



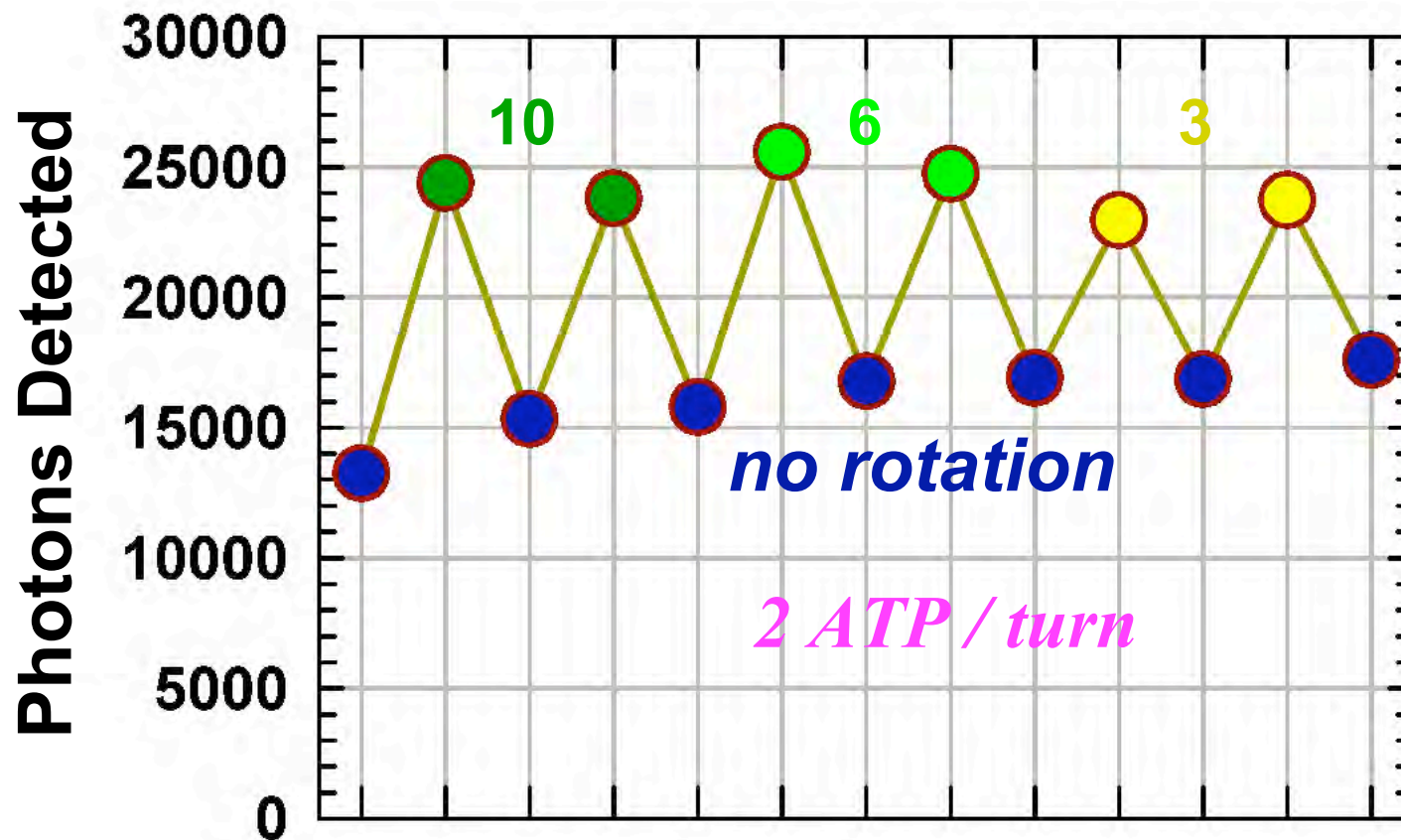
Manipulating Many Motors Simultaneously



Magnet rotated at 10 Hz

Reverse Rotation Led to ATP Synthesis!!

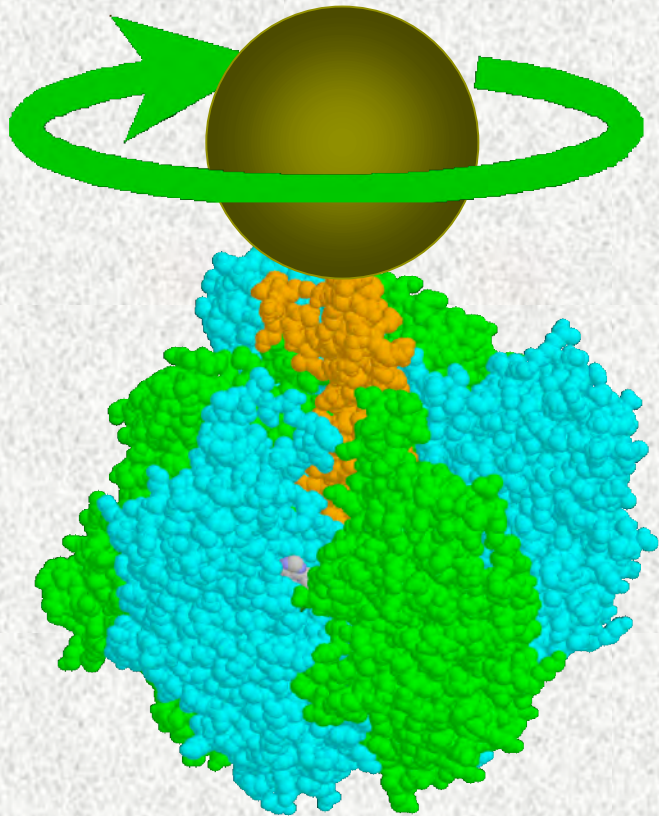
5 min at indicated rates (Hz)



First chemical synthesis driven by force

Hiro Itoh @Hamamatsu Photonics Co., Ltd.

Chemical Synthesis by Force (Torque)!



Free energy of ATP hydrolysis:

$$\Delta G = \Delta G_0 + k_B T \ln [ADP][Pi]/[ATP]$$

$$\Delta G_0 = -58 \text{ pN}\cdot\text{nm} \quad (\text{pH } 7.6)$$

$$\Delta G = - (80 \sim 100) \text{ pN}\cdot\text{nm} \quad (\text{in cells})$$

$$= - (20 \sim 25) k_B T$$

$$\Delta G \sim - 30 \text{ pN}\cdot\text{nm} \quad (\text{this experiment})$$

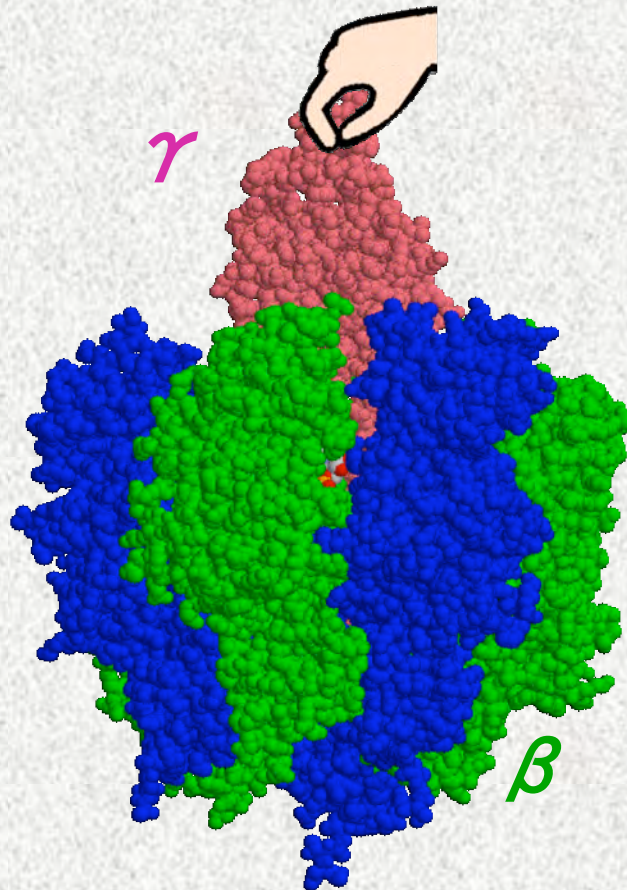
$$\sim - 7 k_B T$$

$$[ADP]=200 \mu\text{M}, [Pi]=10 \text{ mM}, [ATP] \sim 1 \text{ nM}$$

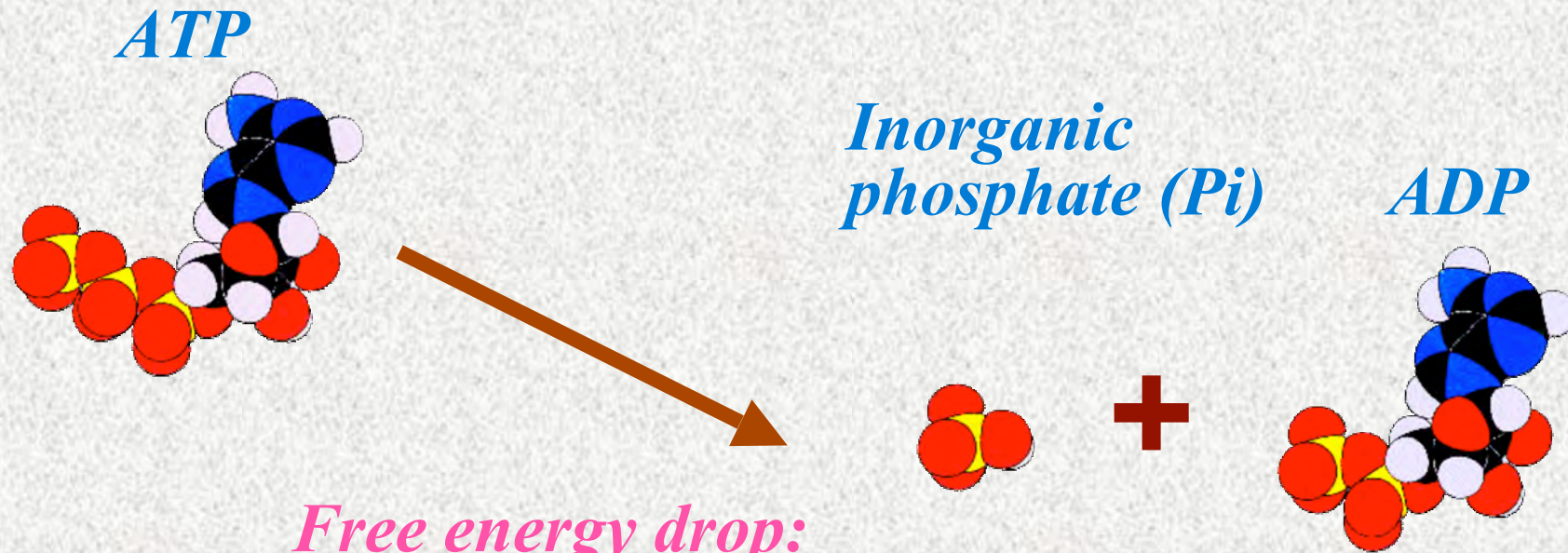
A Force (Torque) Applied to One Point Can Reverse the Operation of F_1 !!

□- Dictator Mechanism:

*The □angle dictates how chemical reactions proceed in □'s.
In hydrolysis, three □'s cooperate through the □angle.*



How Can a Molecular Machine Convert Chemical Energy of ATP Hydrolysis into Mechanical Work?



Free energy drop:

$$\Delta G = \Delta G_0 + k_B T \ln[ADP][Pi]/[ATP]$$

$$\Delta G_0 = -50 \text{ pN}\cdot\text{nm} \quad (@\text{pH } 7)$$

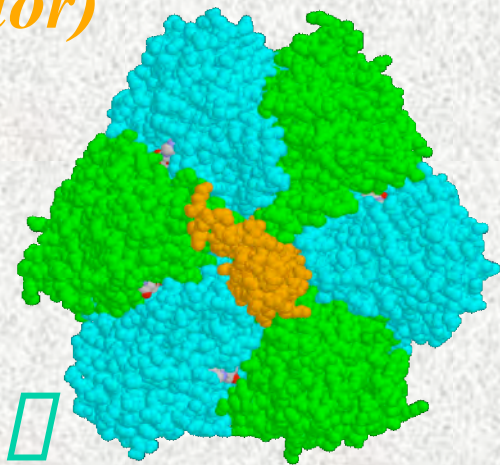
$$\Delta G = -(80\sim 100) \text{ pN}\cdot\text{nm} \text{ in cells}$$

$$= -(20\sim 25) k_B T$$

Liberation of free energy == Reaction goes toward right

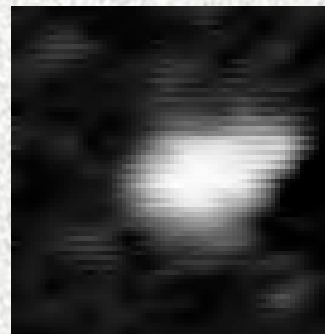
Stepwise Rotation of F_1

\square (rotor)



\square (catalytic)

Watch!!



[ATP] = 20 nM